

Electronic supplementary information

SUPRAMOLECULAR COMPLEXES OF CUCURBITURILS WITH SECOND GROUP METAL SALTS AND HYDRATES AND HISTAMINE

Yu. A. Borisov and S. S. Kiselev*

*Nesmeyanov Institute of Organoelement Compounds, Russian Academy of Sciences,
ul. Vavilova 28, Moscow, 119991 Russia*

Table S1. Results of the computational study of the complexes of cucurbiturils with second group metal salts and hydrates using DFT B3LYP/6-31G* and DFT B3LYP/LanL2DZ methods

Compound	$-E_{\text{tot}}$, a. u.	$-E_{\text{zpve}}$, a. u.	$-E_{\text{G}}$, a. u.	S, cal/K
6-31G(d) basis				
Q8	4814.0026	4812.8901	4812.9983	377.3
Q7	4212.2517	4211.2785	4211.3803	345.3
Q6	3610.4968	3609.6627	3609.7511	298.4
H ₂ O	76.4090	76.3878	76.4055	45.1
CaCl ₂	1598.1063	1598.1044	1598.1316	69.2
Q7_Be ²⁺	4226.5654	4225.5853	4225.6786	326.3
Q7_Be ²⁺ -W ₁	4303.1017	4302.0941	4302.1858	325.2
Q7_Be ²⁺ -W ₂	4379.5820	4378.5482	4378.6405	329.6
Q7_Be ²⁺ -W ₃	4456.0397	4454.9831	4455.0797	344.3
Q7_Be ²⁺ -W ₄	4532.4878	4531.4047	4531.5051	357.3
Q7_Be ²⁺ -W ₅	4608.9369	4607.8236	4607.9231	359.1
Q7_Be ²⁺ -W ₆	4685.3399	4684.2042	4684.3071	371.7
Q7_Be ²⁺ -W ₇	4761.7481	4760.5899	4760.6934	377.7
Q7_Be ²⁺ -W ₈	4838.2909	4837.0997	4837.2037	381.1
Q6_Be ²⁺	3624.8096	3623.9674	3624.0451	271.7
Q6_2Be ²⁺	3638.7663	3637.9187	3637.9953	270.7
Q6_CaCl ₂	5208.6796	5207.8398	5207.9290	309.8
Q6_2CaCl ₂ (I)	6806.8580	6806.0130	6806.1071	330.6
Q6_2CaCl ₂ (II)	6806.8874	6806.0427	6806.1358	328.1
LanL2DZ basis				
Q8	4813.1127	4812.0061	4812.1085	363.0
Q7	4211.4760	4210.5076	4210.6000	323.3
Q6	3609.8342	3609.0039	3609.0852	280.9
Q7_Ba ²⁺	4236.4445	4235.4719	4235.5630	323.3
Ba ²⁺	24.6518	–	–	–
H ₂ O	76.4143	76.3936	76.4112	45.1
BaI ₂	48.1340	48.1334	48.1673	84.9
Q7_Ba ²⁺ -W ₁	4312.9249	4311.9260	4312.0219	338.7
Q7_Ba ²⁺ -W ₂	4389.3976	4388.3716	4388.4678	342.5
Q7_Ba ²⁺ -W ₃	4465.8547	4464.8023	4464.8989	346.2
Q7_Ba ²⁺ -W ₄	4542.3163	4541.2384	4541.3382	359.8
Q7_Ba ²⁺ -W ₅	4618.7620	4617.6597	4617.7624	370.3
Q7_Ba ²⁺ -W ₆	4695.2182	4694.0894	4694.1945	381.2
Q7_Ba ²⁺ -W ₇	4771.6720	4770.5156	4770.6194	380.7
Q7_Ba ²⁺ -W ₈	4848.1031	4846.9228	4847.0293	389.8

Q6_Ba ²⁺	3634.8274	3633.9933	3634.0784	292.6
Q6_2Ba ²⁺	3659.5186	3658.6808	3658.7638	289.5
Q7_BaI ₂	4259.7200	4258.7481	4258.8510	358.9
Q6_BaI ₂	3658.0953	3657.2614	3657.3530	316.0
Q6_2BaI ₂ (I)	3706.2895	3705.4539	3705.5579	349.4
Q6_2BaI ₂ (II)	3706.3649	3705.5266	3705.6306	355.5
Q6_BaCl ₂	3665.2206	3664.3859	3664.4723	304.8
Q6_2BaCl ₂	3720.5644	3719.7268	3719.8238	337.0
Q6_CaI ₂	3669.3995	3668.5643	3668.6537	310.2
Q6_2CaI ₂	3728.9237	3728.0871	3728.1865	340.2
Q8_BaI ₂	4861.3455	4860.2358	4860.3493	400.5

Table S2. Results of the computational study of the complexes of Q7 with HA and Ca²⁺ in water at the B3LYP-D3 or wB97xD (in parentheses) level using the 6-31G* basis and SMD solvation method

Compound	-E _{tot} , a. u.	-E _{zpve} , a. u.	-E _G , a. u.	S, cal/K
Q7-HA	4573.1268 (4571.5044)	4571.9946 (4570.3562)	4572.0946 (4570.4542)	356.1 (349.5)
Q7-W ₈	4824.4202 (4822.7276)	4823.1559 (4821.5250)	4823.3392 (4821.6292)	385.6 (380.5)
Q7	4212.8383 (4211.354)	4211.8585 (4210.3629)	4211.9500 (4210.4494)	320.6 (306.5)
HA	360.2341 (360.07893)	360.0792 (359.9307)	360.1206 (359.9630)	87.1 (85.7)
W ₈	611.5033 (611.3089)	611.2857 (611.1065)	611.3506 (611.1515)	136.8 (135.5)
Q7-HA-2Ca ²⁺	5928.1880	5926.9781	5927.1487	359.029
Q7-HA-H ⁺	4573.6205	4572.4040	4572.5729	355.6
Q7-HA-2H ⁺	4574.0896	4572.928	4573.0266	352.1
Q7-HA-H ⁺ -Ca ²⁺	5251.1466	5249.9976	5250.0976	357.4
Q7-HA-2H ⁺ -Ca ²⁺	5251.6126	5250.3784	5250.5478	356.5
H ₃ O ⁺	76.8362	76.803572	76.8220	46.8
H ₂ O	76.4227	76.401707	76.4194	45.2
W ₆	458.6179	458.4689	458.5072	111.7
Ca ²⁺ -W ₆	1136.1240	1135.9786	1136.0216	127.4

Table S3. Results of the computational study of the complexes of Q7 with HA, 2Ca²⁺, and water in the explicit form at the wB97xD level using the 6-31G* basis

	Reaction	ΔG, kcal/mol	ΔH, kcal/mol
1	Q7-HA-2Ca ²⁺ + W → Q7-HA-2Ca ²⁺ -W ₁	-16.1	-26.9
2	Q7-HA-2Ca ²⁺ -W ₁ + W → Q7-HA-2Ca ²⁺ -W ₂	-13.3	-26.2
3	Q7-HA-2Ca ²⁺ -W ₂ + W → Q7-HA-2Ca ²⁺ -W ₃	-2.8	-15.4
4	Q7-HA-2Ca ²⁺ -W ₃ + W → Q7-HA-2Ca ²⁺ -W ₄	-0.1	-10.5

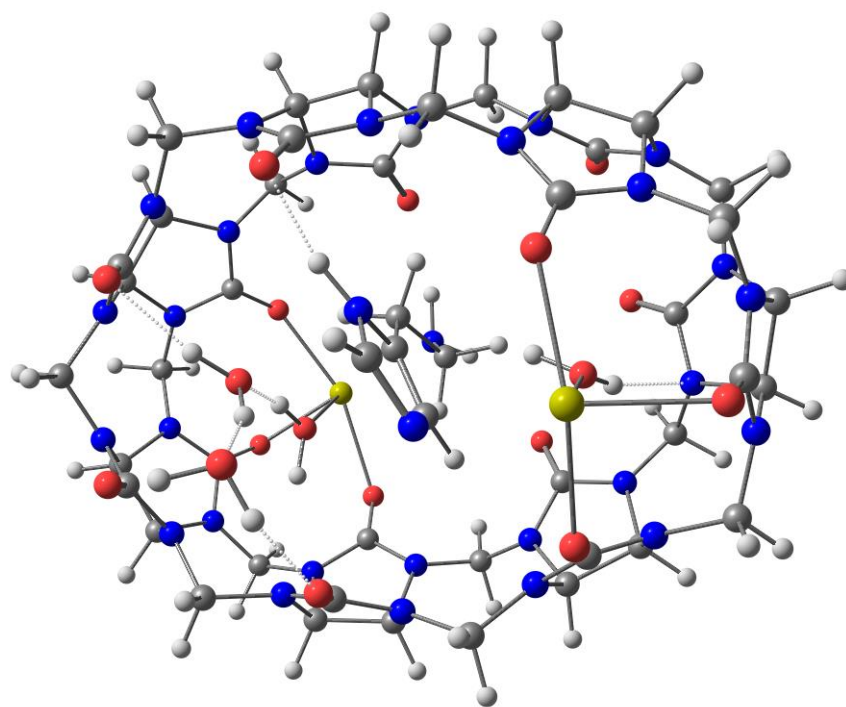


Fig. S1. Structure of complex Q7-HA-2Ca²⁺-W₄.